

National Institute for **Public Health and** the Environment



Datasheet 1-Desoxycarbadox

Reference number : CEC/MAT : 20

Date of preparation : 1994.06.02

date : 1998.01.06

source : CSL

"Bank of Reference Standards"

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Contract number		: MAT-CT92-0020[388710]				
Reference number		: CEC/MAT 20				
Last update Quantity	:	1998 1.54	8.01.06 mg	Chemical Purity	:	>95 %

Figure 1.Molecular structure of 1-desoxycarbadox

Name: Methyl 3-(2-quinoxalinylmethylene)-carbazate N4-oxideSynonym: 1-DesoxycarbadoxMolecular formula: $C_{11}H_{10}N_4O_3$ Molecular weight: 246.226

Long term stability tested on 1997.11.10 : 97.7 ± 0.8 % (storage 4 °C, analysis HPLC-UV, 6 tests on 2 ampoules)

Methods of Characterization:

- I UV spectroscopy
- II IR spectroscopy
- III Mass spectroscopy
- IV ¹H-NMR spectroscopy

I UV Spectroscopy

Instrument: Hitachi U 3000 Method: Dissolved in ethanol (20mg/l)



Figure 2. The UV spectrum of 1-desoxycarbadox

Peak no.	wavelength (nm)	absorbance
1	373.00	0.3340
2	284.00	1.8784
3	205.00	0.8443

II IR-Spectroscopy

Instrument: Perkin Elmer STIR 1720X Sampling technique: nujol mull



Figure 3. The IR-spectrum of 1-desoxycarbadox

Wavelength (cm-1)	designation
1734	C=O stretch
1576	C=C stretch in aromatic compounds
1526	
1494	C=C stretch in aromatic compounds
1248	
1164	C-O stretch
1046	
794	C-H out of plane deformations
768	C-H out of plane deformations

III Mass spectroscopy

Instrument: Kratos MS 25 Method: Direct probe, 70 ev electron impact



Figure 4. The mass spectrum of 1-desoxycarbadox

m/u	percentage	designation
247	4	M + 1
246	31	Μ
230	21	M - O
214	4	M - CH ₃ OH
187	100	M - CH ₃ OCO
171	71	$M - (CH_3OCO + O)$
143	73	$M - (NNCH_3OCO + O)$

III ¹H-NMR Spectroscopy

Instrument: GX 400 Solvent: DMSO- d_6 with TMS (d = 0.0) as internal standard.



Figure 5. The NMR spectrum of 1-desoxycarbadox

Chemical Shift (d)	number of protons	designation
3.77	3	CH ₃
7.84	1	H6 or H7 (arom.)
7.94	1	H6 or H7 (arom.)
8.11	1	H11
8.13	1	H5 or H8 (arom.)
8.44	1	H5 or H8 (arom.)
8.72	1	H3 (arom.)
11.76	1	NH

Preparation and validation of reference standards

The production of ampoules containing the reference material was described in the final report produced September 1995.

The thermal stability of the compound was under investigation and the results for

1-desoxycarbadox over a period of one year are expressed in the table and chart below.



1-Desoxycarbadox Stability Trials

Results of 1-desoxycarbadox stability trials

The results below are the % recovery (with cv) of 1-desoxycarbadox at 4 different storage temperatures over a period of 12 months compared to a standard equivalent to 2 mg.

	temp.	t = 0 months	t = 2 months	t = 6 months	t = 12 months
	(°C)	(% Recovery)	(% Recovery)	(% Recovery)	(% Recovery)
1-DSC	- 20°C + 4°C Room Temp. + 37°C	77.0 +/- 1.5	75.5 +/- 4.0 77.5 +/- 2.5 76.5 +/- 4.5 75.5 +/- 1.5	77.0 +/- 2.5 76.0 +/- 2.0 77.0 +/- 2.5 76.0 +/- 1.5	74.5 +/- 1.5 75.0 +/- 3.5 75.0 +/- 3.5 73.5 +/- 3.0

Conclusion

The spectroscopic data is consistant with the proposed structure for all the methods of determination although a small amount of DMF (which was used as solvent during ampouling) was detected by all the methods of characterization* except UV spectroscopy. No significant impurities were detected by any of the methods of characterization employed.

The results from the stability trials indicate that 1-desoxycarbadox is acceptably stable over a period of one year at temperatures up to $37 \,^{\circ}$ C.

* At approximately 2.8 and 2.9 d in the NMR spectrum, 1680 cm⁻¹ in the IR spectrum and 73 m/u in the mass spectrum.